

ALS

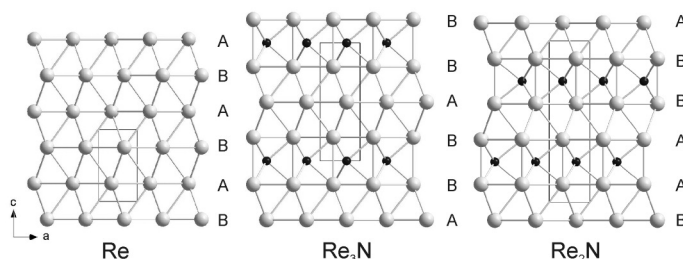
SCIENCE HIGHLIGHT

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## Two Novel Ultra-Incompressible Materials

Some current challenges in aerospace engineering and fission/fusion applications require materials that are mechanically and chemically stable at extreme conditions. One such class of materials is ultrahigh-temperature ceramics, which are often binary transition-metal carbides, borides, or nitrides. It is therefore of great interest to understand how to synthesize new compounds of this type. A research team from Germany, the United Kingdom, and Berkeley Lab working at ALS Beamlines 12.2.2 and 12.3.2 has now synthesized and characterized two novel bulk rhenium nitrides,  $\text{Re}_2\text{N}$  and  $\text{Re}_3\text{N}$ . Both phases are extremely incompressible, and  $\text{Re}_3\text{N}$  is also better placed for potential technological applications than are other incompressible transition-metal carbides and nitrides of the period-six elements because it can be formed at relatively moderate pressures and temperatures.

The introduction of smaller atoms such as boron, nitrogen, or carbon into interstitial sites in close-packed transition-metal lattices leads to dramatic changes in the physical properties in the compound with respect to that of the metal. For example, the incorporation of carbon into group IV or V transition-metals leads to an increase in melting temperature by 1000–1500 K, yielding binary transition-metal carbides



**The incorporation of nitrogen (black balls) into the rhenium (light grey) hcp lattice (AB stacking sequence of rhenium atoms) at increasing pressure and temperature conditions leads to the formation of  $\text{Re}_3\text{N}$  (ABB stacking) and  $\text{Re}_2\text{N}$  (AABB sequence).**

**Nitrogen atoms occupy interstitial sites between AA or BB layers only.**

with extremely high melting points. This characteristic, in addition to high hardness and incompressibility, is required for materials used in abrasive or coating applications.

The origin of the unusual properties is the complex bonding found in these compounds, where there are metal-metal, metal-nonmetal, and nonmetal-nonmetal contacts. Combining metals with high densities of valence electrons (such as rhenium, osmium, and iridium) with light elements is especially promising; however, extreme pressure and temperature conditions are required to induce reactions.

In order to synthesize rhenium nitrides, rhenium foil was pressurized in the range from 10–27 GPa together with nitrogen within a diamond-anvil pressure cell and then heated by two laser beams, one from

each side, up to 2800 K.  $\text{Re}_3\text{N}$  can be synthesized between 10.5–16 GPa and 1700–2250(150) K, and  $\text{Re}_2\text{N}$  was obtained at 20(2) GPa and  $\sim 2000$  K. Reaction products were analyzed by powder x-ray diffraction, in combination with density functional theory calculations. Structural models were proposed and further confirmed by x-ray Laue microdiffraction, allowing for detailed analysis of phase distribution and grain sizes within the samples.

By combining a heavy transition metal with a light, covalent-bond-forming element and adding pressure and temperature, researchers synthesized two novel materials with both high incompressibility and proposed high hardness. An incompressible material is difficult to compress elastically. It is resistant to volume and/or linear compression. A hard mate-

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Research conducted by A. Friedrich, B. Winkler, L. Bayarjargal, and W. Morgenroth (Geowissenschaften, Goethe-Universität, Germany), E.A. Juarez-Arellano (Universidad del Papaloapan, Mexico), V. Milman (Accelrys, UK), K. Refson (Rutherford-Appleton Laboratory, UK), and M. Kunz and K. Chen (ALS).

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rial resists plastic (as opposed to elastic) deformation, which involves irreversible motion of the atoms with respect to each other.

Interesting results were obtained on the crystal chemistry of period-six transition-metal nitrides and carbides. Researchers found that nitrogen dissociates

## Seeking the Ultrahard and Incompressible

Materials that combine mechanical properties (ultrahardness, ultra-incompressibility), thermal properties (ultrahigh melting point), and chemical resistance are useful for numerous industrial applications, such as cutting and grinding tools, abrasives, components of gas turbines, coatings for high-speed drill bits, and electronic and semiconductor components.

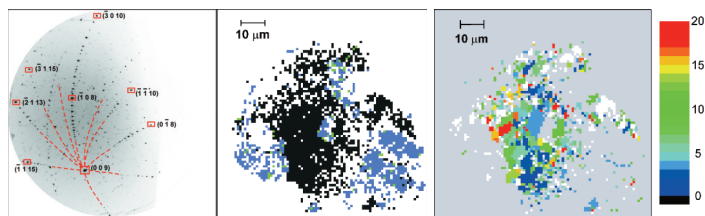
The hardest and most well-known such material is diamond. Because it occurs naturally, it has become widely used, but even diamond has limitations. For example, it is not effective for cutting material made of iron. Though much effort has been exerted to find affordable synthesis routes for diamond, it generally requires high pressure and temperature, as do most materials of this kind, making it expensive. So, the search for new ultrahard materials that combine these useful mechanical, thermal, and chemical properties has intensified.

Future research will endeavor to further characterize the properties of these new materials for industrial applications, and to synthesize them in larger quantities and at lower cost.

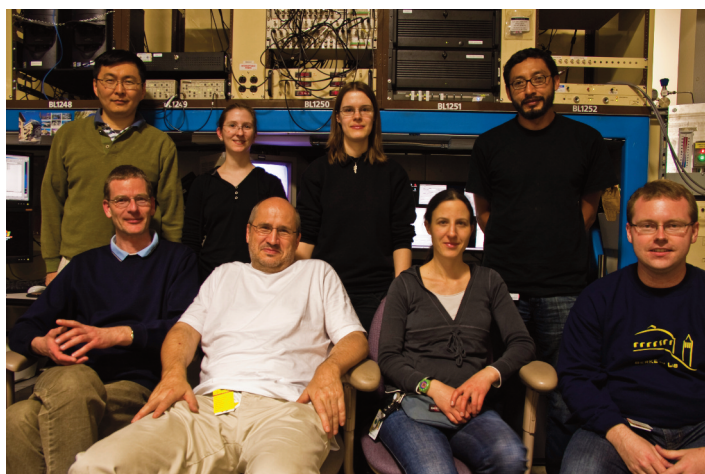
during  $\text{Re}_3\text{N}$  and  $\text{Re}_2\text{N}$  synthesis, similar to the phase formation of  $\text{TiN}$ ,  $\text{TaN}$ , and  $\text{Ta}_2\text{N}_3$  from the elements. This is in contrast to other known transition-metal nitrides of period-six elements with higher atomic numbers, i.e.,  $\text{OsN}_2$ ,  $\text{IrN}_2$ , and  $\text{PtN}_2$ . Close structural relationships are observed with rhenium carbide,  $\text{Re}_2\text{C}$ , which was also synthesized at high pressures and temperatures.

$\text{Re}_3\text{N}$  and  $\text{Re}_2\text{N}$  were char-

acterized to be ultra-incompressible, with bulk moduli of  $>400$  GPa, similar to the most incompressible binary transition-metal carbides and nitrides found to date and significantly less compressible than pure rhenium. The rhenium nitrides synthesized here are potential candidates as ultrahard materials and may find some special applications in electron conductivity at extremely high temperatures and pressures.



**Typical results from white-beam x-ray microdiffraction. Left: An image indexed with the unit cell of  $\text{Re}_3\text{N}$ . The indices of a few reflections are shown as examples. The red dotted curves indicate extensions of the indexed Laue zones. Middle and right: A map of the phase distribution ( $\text{Re}_3\text{N}$  and  $\text{Re}$  in black and blue areas, respectively) and the distribution of the  $\text{Re}_3\text{N}$  grains and their sizes ( $\sim 3\text{--}8$   $\mu\text{m}$ ) derived from the c-axis orientation of  $\text{Re}_3\text{N}$ .**



**Bottom row: Wolfgang Morgenroth, Björn Winkler, Alexandra Friedrich, and Florian Schröder. Top row: Lkhamsuren Bayarjargal, Jasmin Biehler, Nadine Rademacher, and Erick A. Juarez-Arellano.**